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# Chemical constituents produced by the marine fungi *Penicillium* sp. M839 and their antimicrobial evaluation

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#### **ABSTRACT**

From the agar-based culture of the marine-derived *Penicillium* sp. M839 strain, nine known compounds were isolated and structurally determined, including fumiquinazoline D (1), fumiquinazoline B (2), cerevisterol (3), stoloniferol B (4), norhaman (5), 3,4-dihydroxy-6,7-dimethyl-quinolin-2-carboxylic (6), uracil (7), 3-methyl uracil (8), thymine (9). These compounds were characterized via 1D and 2D NMR spectroscopic and mass spectrometric analyses. Compounds 2 and 8 were first recognized from the genus *Penicillium*, while the remaining compounds were previously isolated from this genus. Compounds 1–6 were evaluated for their antimicrobial activity against a panel of test microorganisms (three Gram-positive bacteria: *Enterococcus faecalis, Staphylococcus aureus, Bacillus cereus*; three Gram-negative bacteria: *Escherichia coli, Pseudomonas aeruginosa, Salmonella enterica* and one yeast strain *Candida albicans*). The results showed that compound 6 displayed strong antimicrobial activity against two Gramnegative bacteria *E. coli* and *S. enterica*, with the MIC values of 32 and 64 μg/mL, respectively. In addition, the remaining compounds 1–5 inhibited against all test microorganisms with MIC values ranging from 64 to 256 μg/mL.

Keywords: Penicillium sp. M839, marine fungi, alkaloid, sterol, antimicrobial.

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#### INTRODUCTION

Penicillium has gained significant recognition as one of the most well-known fungal genera due to the discovery of bioactive compounds [1, 2]. This genus is ubiquitous, found in diverse environments, and is especially prevalent in marine ecosystems, where it is one of the most widespread fungal groups [1, 2]. Notably, various Penicillium species have already led to the development of major clinical drugs [2]. It is impossible to discuss Penicillium without mentioning penicillin, а discovery revolutionized medical treatment by introducing antibiotics and eradicating numerous human diseases caused by pathogenic bacteria [2]. Since terrestrial *Penicillium* species are known to produce many biologically active compounds, their abundance in marine environments (marine-derived fungi from the *Penicillium* genus) presents a promising source of novel agents with potential therapeutic applications [2, 3]. In Vietnam, there are limited publications on the chemical constituents of the marine fungi Penicillium [4]. During our screening program, the EtOAc extract of the Penicillium sp. M839 strain exhibited antimicrobial activity against two Gram-positive bacteria and one Gramnegative bacteria strain: Staphylococus aureus ATCC25923, Enterococcus faecalis ATCC29212, Escherichia coli ATCC25922 with MIC values of 32, 128 and 64  $\mu$ g/mL, respectively. This paper reported the isolation and structural elucidation of nine compounds 1-9 from the marine-derived fungus Penicillium sp. M839 strain was isolated from sediment samples collected at the Ran Trao conservation area - Khanh Hoa sea area (Vietnam).

#### MATERIALS AND METHODS

# Marine sediment sample

The marine sediment sample (Code: HP03/20-VP04-TT07) was collected in May 2021 at the Ran Trao conservation area, Khanh Hoa sea area, Vietnam. It was collected at 11 m depth with geographic coordination of 12.0°37′34″ - 109.0°12′49″ and water temperature of 28°C.

This sample was collected in a sterile 50 mL Falcon tube, preserved on ice, and processed within 24 hours.

# Isolation and Identification of the fungus M839

The sediment sample was homogenized and subjected to a wet-heat treatment at  $60^{\circ}$ C for 6 minutes. The suspension was diluted using a ten-fold dilution series down to  $10^{-3}$ . Aliquots of 50  $\mu$ L were then spread onto Petri dishes filled with PDA solid medium (30 g/L potato extract, 20 g/L dextrose, 30 g/L Instant Ocean, and 15 g/L agar). The plates were incubated at 28°C for 7 days. A colony of the fungus M839 was subsequently transferred to a fresh Petri dish containing PDA medium for purification (Fig. 1).

The strain M839 was identified as belonging to the genus *Penicillium* through 18S rRNA gene sequence analysis, which was compared with fungal sequences in the GenBank database using the NCBI BLAST program.



Figure 1. Morphological appearance of M839 strain's colonies

### Fermentation fungus M839

The strain M839 was activated and inoculated into 1 L of PDB broth medium pH 7.0 (comprising 30 g/L potato extract, 20 g/L

dextrose, 30 g/L instant ocean). After 7 days of incubation at 28°C with shaking at 100 rpm, the culture broth was spread on the medium surface of 50 flasks (each 3 L flask containing 1 L of PDA medium, pH 7.0). The flasks were incubated at 28°C and harvested on the twenty-one days.

# General experiment procedures

Optical rotations were recorded on a JASCO P-2000 polarimeter using a sodium (589 nm, D line) lamp. Melting points were measured using a Thermo Mel-Temp 3.0 (Thermo Scientific). The NMR spectra were recorded on Bruker Avance 500 MHz and 600 MHz spectrometers, and tetramethylsilane (TMS) was used as an internal standard. Thin layer chromatography used pre-coated silica gel Merck 60  $F_{254}$ . Column chromatography (CC) was performed on silica gel (Kiesel gel 60, 70–230 mesh, and 230–400 mesh), or Sephadex LH-20 resins (Sigma-Aldrich, USA), and reversed-phase  $C_{18}$  (RP-18, 30–50  $\mu$ m). All solvents used in column chromatography were distilled before use.

# Extraction and isolation

The culture of *Penicillium* sp. M839 (45 kg) were cut into small pieces and sonicated in ethyl acetate (30 L) at 45-50°C (5 times for 4 hours each). The filtrated solution was concentrated under reduced pressure to give the crude ethyl acetate extract (65.4 g). The EtOAc extract was subjected to a silica gel column chromatography (CC) and eluted with a solvent gradient mixture CH<sub>2</sub>Cl<sub>2</sub>/MeOH to yield six fractions F1-F6. Purification of fraction F2 (5.3 g) by a Sephadex LH-20 column using MeOH/CH<sub>2</sub>Cl<sub>2</sub> (90/1, v/v), followed by a silica gel CC using  $CH_2Cl_2$ /acetone (5/1, 4/1, 3/1, v/v), afforded compound 4 (13 mg) and compound 5 (8.6 mg). Fraction F3 (534 mg) was subjected to CC on Sephadex LH-20 (CH<sub>2</sub>Cl<sub>2</sub>/MeOH (2/8, v/v)), followed by CC on a silica gel column (dichloromethane/ acetone (4/1, 3/1, v/v) to afford compound 8 (16 mg). Fraction F4 (2.1 mg) was separated into six subfractions (F4.1-F.4.6) by CC on silica gel using the solvent mixtures of CH<sub>2</sub>Cl<sub>2</sub>/MeOH gradient. Subfractions F4.3 (123 mg) and F4.4 (121 mg) were recrystallized in a solvent mixture of  $CH_2Cl_2/MeOH$  (9/1, v/v) to afford compound 9 (41 mg) and compound 7 (21 mg), respectively. Fraction F5 (4.7 g) was chromatographed on a reversed-phase C<sub>18</sub> (RP- C<sub>18</sub>) silica gel column using an  $H_2O/MeOH$  solvent system (2/1, 1/1, v/v) to obtain two subfractions (F5.1 and F5.2). Compounds 3 (7.8 mg) and 6 (29 mg) were purified from sub-fraction F5.1 (643 mg) by subjecting them to silica gel CC using a CH<sub>2</sub>Cl<sub>2</sub>/MeOH gradient. Compounds **1** (11 mg) and 2 (9 mg) were purified from sub-fraction F5.2 (1.3 g) after being subjected to a Sephadex LH-20 column with MeOH/CH<sub>2</sub>Cl<sub>2</sub> (8/2, v/v), followed by a silica gel CC using CH<sub>2</sub>Cl<sub>2</sub>/acetone gradient.

Fumiquinazoline D (1): White solid, mp: 215–216°C,  $\left[\alpha\right]_{D}^{25}$  = + 56.7 (*c* 1.73, CHCl<sub>3</sub>). <sup>1</sup>H-NMR (acetone- $d_6$ , 600 MHz):  $\delta_{H}$  (ppm) 1.13 (3H, d, J = 6.6 Hz, CH<sub>3</sub>-29), 2.18 (3H, s, CH<sub>3</sub>-16),2.32 (1H, dd, J = 15.0, 0.6 Hz,  $H_a$ -15), 3.21 (1H, dd, J = 15.0, 10.4 Hz, H<sub>b</sub>-15), 4.25 (1H, td, J =6.6, 1.2 Hz, H-20), 5.42 (1H, dd, J = 9.0, 0.2 Hz, H-14), 5.61 (OH-17), 5.76 (1H, d, J = 1.8 Hz, H-18), 7.16 (1H, td, J = 7.8, 1.2 Hz, H-26), 7.30 (1H, td, J = 7.8, 1.2 Hz, H-25), 7.43 (1H, d, J =7.8 Hz, H-24), 7.52 (1H, d, J = 7.2 Hz, H-27), 7.54 (1H, td, J = 7.8, 1.2 Hz, H-9), 7.72 (1H, d, J= 7.8 Hz, H--7, 7.83 (1H, td, J = 7.8, 1.8 Hz, H--8),8.18 (1H, dd, J = 7.8, 1.2 Hz, H-10). <sup>13</sup>C-NMR (acetone- $d_6$ , 150 MHz):  $\delta_{\mathbb{C}}$  (ppm) 17.9 (C-29), 19.3 (C-16), 44.1 (C-15), 54.0 (C-14), 59.5 (C-20), 71.9 (C-3), 84.4 (C-17), 86.5 (C-18), 115.8 (C-24), 121.7 (C-11), 125.1 (C-27), 125.9 (C-26), 127.4 (C-10), 128.0 (C-9), 128.5 (C-7), 130.1 (C-25), 135.3 (C-8), 139.1 (C-23), 139.7 (C-28), 147.6 (C-6), 154.4 (C-4), 161.2 (C-12), 169.6 (C-1), 172.9 (C-21).

Fumiquinazoline **B** (2): Pale yellow solid, mp: 174–175°C,  $\left[\alpha\right]_{D}^{25}$  = -176.8 (*c* 0.23, CHCl<sub>3</sub>), <sup>1</sup>H-NMR (acetone-*d*<sub>6</sub>, 600 MHz): δ<sub>H</sub> (ppm) 1.23 (3H, d, *J* = 6.6 Hz, CH<sub>3</sub>-29), 1.87 (3H, d, *J* = 7.2 Hz, CH<sub>3</sub>-16), 2.47 (1H, dd, *J* = 13.8, 5.4 Hz, H<sub>3</sub>-15), 2.64 (1H, dd, *J* = 13.8, 11.4 Hz, H<sub>b</sub>-15), 4.17 (1H, td, *J* = 6.6, 1.2 Hz, H-20), 4.81 (1H, q, *J* = 14.4, 7.2 Hz, H-3), 5.36 (1H, d, *J* = 5.4 Hz, H-18), 5.70 (1H, td, *J* = 7.8, 1.2 Hz, H-26), 7.33 (1H, td, *J* = 7.8, 1.2 Hz, H-25), 7.48 (1H, d, *J* = 9.0 Hz, H-24), 7.51 (1H, td, *J* = 7.2, 1.2 Hz, H-9), 7.62 (1H,

d, J = 7.8 Hz, H-7), 7.63 (1H, d, J = 7.2 Hz, H-27), 7.81 (1H, td, J = 7.2, 1.8 Hz, H-8), 8.15 (1H, dd, J = 7.8, 1.2 Hz, H-10). CNMR (acetone- $d_6$ ,150 MHz):  $\delta_{\rm C}$  (ppm) 18.4 (C-29), 24.9 (C-16), 39.8 (C-15), 52.9 (C-14), 53.4 (C-3), 59.5 (C-20), 80.9 (C-17), 87.4 (C-18), 115.2 (C-24), 121.2 (C-11), 125.5 (C-27), 126.2 (C-26), 127.3 (C-7), 127.5 (C-10), 127.8 (C-9), 129.9 (C-25), 135.4 (C-8), 138.2 (C-23), 140.5 (C-28), 148.4 (C-6), 153.0 (C-4), 161.0 (C-12), 171.0 (C-1), 171.7 (C-21).

Cerevisterol (3): White solid, mp: 225-226°C,  $\left[\alpha\right]_{D}^{23} = -67 \text{ (0.07, MeOH), }^{1}\text{H-NMR}$ (DMSO- $d_6$ , 600 MHz):  $\delta_H$  (ppm) 0.54 (3H, s, CH<sub>3</sub>-18), 0.80 (3H, d, J = 7.2 Hz, CH<sub>3</sub>-27), 0.82 (3H, d, J = 7.8 Hz, CH<sub>3</sub>-26), 0.88 (3H, d, J = 6.6 Hz, CH<sub>3</sub>-28), 0.90 (3H, s,  $CH_3$ -19), 0.99 (3H, d, J = 6.6 Hz,  $CH_3$ -21), 3.36 (1H, d, J = 5.4 Hz, H-6), 3.58 (1H, s, OH-5), 3.77 (1H, m, H-3), 4.23 (1H, d, J = 5.4Hz, OH-3), 4.48 (1H, d, J = 5.4 Hz, OH-6), 5.05 (1H, dd, J = 2.4, 5.4 Hz, H-7), 5.17 (1H, dd, J =8.4, 15.6 Hz, H-22), 5.24 (1H, dd, J = 7.2, 15.6 Hz, H-23). <sup>13</sup>C-NMR (DMSO- $d_6$ , 150 MHz):  $\delta_{\rm C}$ (ppm) 12.1 (C-18), 17.3 (C-28), 17.7 (C-19), 19.5 (C-26), 19.7 (C-27), 21.0 (C-21), 21.3 (C-11), 22.6 (C-15), 27.7 (C-16), 31.2 (C-2), 32.5 (C-1, C-25), 36.7 (C-10), 39.2 (C-12), 39.6 (C-4), 40.2 (C-20), 42.0 (C-24), 42.3 (C-9), 43.0 (C-13), 54.2 (C-14), 55.3 (C-17), 66.0 (C-3), 72.1 (C-6), 74.5 (C-5), 119.4 (C-7), 131.4 (C-23), 135.4 (C-22), 139.7 (C-8).

Stoloniferol B (4): Colorless needle, mp:  $140-141^{\circ}$ C,  $\alpha$ <sub>D</sub><sup>25</sup> = +93.3 (c 0.05, CHCl<sub>3</sub>), <sup>1</sup>H-NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$ <sub>H</sub> (ppm) 1.31 (3H, d, J = 6.6 Hz, CH<sub>3</sub>-13), 1.33 (3H, d, J = 6.6 Hz, CH<sub>3</sub>-12), 2.10 (3H, s, CH<sub>3</sub>-11), 2.97 (1H, dt, J = 7.2, 6.0 Hz, H-4), 4.68 (1H, dt, J = 7.8, 6.6 Hz, H-3), 6.30 (1H, s, H-7), 11.35 (1H, s, OH-8). <sup>13</sup>C-NMR (CDCl<sub>3</sub>,150 MHz):  $\delta$ <sub>C</sub> (ppm) 9.9 (C-11), 19.7 (C-12), 20.0 (C-13), 34.8 (C-4), 80.0 (C-3), 100.4 (C-9), 101.4 (C-7), 113.3 (C-5), 143.1 (C-10), 161.0 (C-6), 162.4 (C-8), 168.6 (C-1).

Norhaman (5): White solid, mp:  $198-200^{\circ}$ C,  $^{1}$ H NMR (500 MHz, CD<sub>3</sub>OD):  $\delta_{\rm H}$  (ppm) 7.29 (1H, dd, J=8.0, 2.5 Hz, H-6), 7.58 (1H, dd, J=8.0, 0.5 Hz, H-7), 7.59 (1H, d, J=8.0 Hz, H-8), 8.11 (1H, d, J=5.0 Hz, H-4), 8.21 (1H, d, J=8.0 Hz, H-5), 8.32 (1H, d, J=5.5 Hz, H-3), 8.81 (1H, s, H-1).  $^{13}$ C NMR (125 MHz, CD<sub>3</sub>OD):  $\delta_{\rm C}$  (ppm) 112.9 (C-8), 116.3 (C-4), 120.7 (C-6), 122.4 (C-4b),

122.5 (C-5), 129.7 (C-7), 130.2 (C-4a), 134.2 (C-1), 137.7 (C-9a), 138.5 (C-3), 142.8 (C-8a).

**3,4-Dihydroxy-6,7-dimethyl** -quinolin-2-carboxylic (6): Pale yellow solid, mp: 154–155°C,  $^1$ H-NMR (CDCl $_3$ +CD $_3$ OD, 600 MHz):  $\delta_{\rm C}$  (ppm) 2.48 (3H, s, C-10), 2.51 (3H, s, C-11), 7.69 (1H, s, H-8); 7.96 (1H, s, H-5). $^{13}$ C-NMR (CDCl $_3$ +CD $_3$ OD, 150 MHz):  $\delta_{\rm C}$  (ppm) 19.2 (C-10), 20.3 (C-11), 125.7 (C-8), 128.5 (C-5), 129.7 (C-4a), 138.6 (C-8a), 139.1 (C-7), 141.5 (C-4), 144.7 (C-6), 146.2 (C-3), 149.8 (C-2), 160.6 (C=O).

**Uracil (7)**: White solid, mp: 155–156°C, <sup>1</sup>H-NMR (500 MHz, DMSO- $d_6$ ):  $\delta_H$  (ppm) 5.46 (1H, d, J = 7.6 Hz, H-5); 7.35 (1H, d, J = 7.5 Hz, H-6).

**3-methyluracil (8)**: White solid, mp: 155–156°C, <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD):  $\delta_{\rm H}$  3.22 (3H, s, CH<sub>3</sub>), 5.52 (1H, d, J = 8.0 Hz, CH-5), 7.63 (1H, d, J = 8.0 Hz, CH-6).

Thymine (9): White solid, mp: 320–323°C;  $^{1}$ H-NMR (500 MHz, CD<sub>3</sub>OD):  $\delta_{H}$  (ppm) 1.88 (3H, s, CH<sub>3</sub>), 7.13 (1H, s, H-6).

# Evaluating antimicrobial activity of compound 1-6

The antimicrobial activity of compounds 1-6 was evaluated using the serial dilution method described by Andrews (2001) at the Institute of Marine Biochemistry, Vietnam Academy of Science and Technology [4, 5]. The samples diluted in DMSO in decreasing concentrations of 256, 128, 64, 32, 16, 8, 4, and 2 μg/mL. Subsequently, 50 μL of bacterial and yeast suspension at a concentration of 2 × 10<sup>5</sup> CFU/mL was added, and the mixture was incubated at 37°C for 24 hours. The minimum inhibitory concentration (MIC) was determined as the lowest concentration of the sample that completely inhibited microbial growth after 24 hours. Streptomycin and nystatin were positive controls for bacteria and yeast, respectively. Seven tested strains used in this study were provided by the American Type Culture Collection (ATCC), including three Gramnegative strains: Escherichia coli ATCC 25922, **Pseudomonas** aeruginosa **ATCC** 27853, Salmonella enterica ATCC 13076; three Grampositive strains: Enterococcus faecalis ATCC 29212, Staphylococcus aureus ATCC 25923, Bacillus cereus ATCC 14579; and one yeast

strain: *Candida albicans* ATCC 10231. The independent experiments were performed in triplicate.

#### **RESULTS AND DISCUSSION**

From the agar-based culture of the marinederived Penicillium sp. M839 strain, nine known compounds were isolated and structurally determined, including fumiquinazoline D (1), fumiquinazoline B (2), cerevisterol (3), stoloniferol B (4), norhaman (5), 3,4-dihydroxy-6,7-dimethyl-quinolin-2-carboxylic (6), uracil (7), 3-methyl uracil (8), thymine (9). These compounds were characterized via 1D and 2D NMR spectroscopic and mass spectrometric analyses (Fig. 2).

Figure 2. Secondary metabolites 1-9 from Penicillium sp. M839

Compound 1 was obtained as a white solid with a positive specific rotation  $\left[\alpha\right]_{0}^{25}$  = +56.7 (c 1.73, CHCl<sub>3</sub>). The <sup>1</sup>H NMR spectrum with the aid of the HSQC spectrum showed the signals of eight aromatic protons at  $\delta_{\rm H}$  7.16–8.18, two methyl groups at  $\delta_H$  1.13 (d, J = 6.6 Hz, CH<sub>3</sub>-29), 2.18 (s, CH<sub>3</sub>-16); three  $sp^3$  methine groups at  $\delta_H$ 4.25 (td, J = 6.6, 1.2 Hz, H-20), 5.42 (dd, J = 9.0, 0.2 Hz, H-14), 5.76 (d, J = 1.8 Hz, H-18) and two methylene protons at  $\delta_{\rm H}$  2.32 (dd, J = 15.0, 0.6 Hz,  $H_a$ -15) and 3.21 (dd, J = 15.0, 10.4 Hz,  $H_b$ -15). The chemical shifts of the three methine groups indicated their direct connection to nitrogen atoms. Analysis of the <sup>13</sup>C-NMR spectrum with the aid of the HSQC spectrum revealed the presence of 24 carbon atoms, including three carbonyl groups, two methyl groups, one  $sp^3$  methylene carbon, three  $sp^3$ methine carbons, eight sp<sup>2</sup> methine carbons, one quaternary carbon, one oxygen bearing tertiary carbon and five non-protonated carbons. The chemical shifts of the carbonyl groups at  $\delta_{\text{C}}$  161.2 (C-12), 169.6 (C-21), and 172.9 (C-1) are related to the carbonyls of the amide groups. Similarly, the chemical shifts of C-3 ( $\delta_{\rm C}$  71.9) and C-17 ( $\delta_{\rm C}$  84.4) suggested their direct linkage to nitrogen or oxygen atoms. In the COSY spectrum of 1, two orthodisubstituted benzene rings were indicated by the COSY correlations of the spin-spin coupling systems: H-7/H-8/H-9/H-10 and H-24/H-25/H-26/H-27 (Figure 3). The interpretation of the HMBC correlations of H-10 ( $\delta_{\rm H}$  8.18) with C-12  $(\delta_{\rm C} 161.2)$ /C-8  $(\delta_{\rm C} 135.3)$ /C-6  $(\delta_{\rm C} 147.6)$ , CH<sub>3</sub>-16  $(\delta_{\rm H} \, 2.18)$  with C-3  $(\delta_{\rm C} \, 71.9)$ /C-4  $(\delta_{\rm C} \, 154.4)$ , H-14  $(\delta_{\rm H} 5.42)$  with C-12  $(\delta_{\rm C} 161.2)$ /C-4  $(\delta_{\rm C} 154.4)$ /C-1 ( $\delta_{\rm C}$  169.6), along with the reasonable arrangement of two nitrogen atoms (N-13, N-5) confirmed the presence of the 6H-pyrazino[2,1blguinazoline-3,6(4H)-dione moiety (A moiety) in compound 1 (Fig. 3). Similarly, the imidazo[1,2-a]indole moiety (B moiety) was established through the second orthodisubstituted benzene ring (from H-24 to H-27),

the reasonable arrangement of two nitrogen atoms (N-19, N-22), and the long-range correlations in the HMBC spectrum as follows: H-27 ( $\delta_{\rm H}$  7.52) with C-17 ( $\delta_{\rm C}$  84.4)/C-23 ( $\delta_{\rm C}$  139.1)/C-25 ( $\delta_{\rm C}$  130.1), H-24 ( $\delta_{\rm H}$  7.43) with C-28 ( $\delta_{\rm C}$  139.7), H<sub>3</sub>-29 ( $\delta_{\rm H}$  1.13) with C-20 ( $\delta_{\rm C}$  59.5)/C-21 ( $\delta_{\rm C}$  172.9), H-20 ( $\delta_{\rm H}$  4.25) with C-21 ( $\delta_{\rm C}$  172.9) (*Figure 3*). Furthermore, the HMBC spectrum showed the long-range correlations of H-18 ( $\delta_{\rm H}$  5.76) with C-3 ( $\delta_{\rm C}$  71.9), H-14 ( $\delta_{\rm H}$  5.42) with C-15 ( $\delta_{\rm C}$  44.1)/C-17 ( $\delta_{\rm C}$  84.42), and H-15 ( $\delta_{\rm H}$  2.32 and 3.21) with C-17 ( $\delta_{\rm C}$  84.4)/C-1 ( $\delta_{\rm C}$  172.9) were observed. These HMBC correlations confirmed that the A moiety was

connected to the B moiety through a methylene bridge (CH<sub>2</sub>-15) at C-14 and C-17 and the linkage between N-19 and C-3. Thus, the planar structure of compound **1** was determined, as shown in Figure 2. The relative configuration of **1** was elucidated based on the NOESY correlations from H-20 ( $\delta_{\rm H}$  4.25) to H-16 ( $\delta_{\rm H}$  2.18), from H-18 ( $\delta_{\rm H}$  5.76) to OH-17 ( $\delta_{\rm H}$  5.61). Thus, based on the analyses of the NMR spectra, optical rotation value, and comparison with reported data [6, 7], compound **1** was identified as fumiquinazoline D, which was previously isolated from the fungus *Aspergillus fumigatus* (1995) [6].

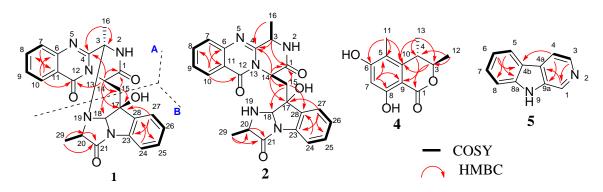


Figure 3. Key HMBC and COSY correlations of compounds 1, 2, 4 and 5

Compound 2 was obtained as a white solid and optically active  $\left[\alpha\right]_{0}^{25} = -176.8$  (c 0.23,  $CHCl_3$ ). The  $^{13}C$  NMR and HSQC spectra of compound 2 displayed the signals of 24 carbons, including three carbonyl groups, two doublet methyl groups, one sp<sup>3</sup> methylene group, four  $sp^3$  methine groups, eight  $sp^4$ methine groups, one oxygen-bearing tertiary carbon and five non-protonated carbons. The <sup>1</sup>H NMR spectrum with the aid of the HSQC and COSY spectra showed the signals of two orthodisubstituted benzene rings (from H-7 to H-10 and from H-24 to H-27), two methyl groups at  $\delta_{\rm H}$  1.23 (d, J = 6.6 Hz, CH<sub>3</sub>-29), 1.87 (d, J = 7.2 Hz, CH<sub>3</sub>-16); two  $sp^3$  methine groups at  $\delta_H$  4.17 (td, J = 6.6, 1.2 Hz, H-20), 4.81 (q, J = 14.4, 7.2)Hz, H-3); one  $sp^3$  methylene at  $\delta_H$  2.47 (dd, J=13.8, 5.4 Hz,  $H_a$ -15), 2.64 (dd, J = 13.8, 11.4 Hz, H<sub>b</sub>-15). Comparison of the 1D and 2D-NMR spectra of compound 2 with those of 1 indicated that compound 2 had the same fumiquinazoline skeleton as compound 1, except the presence of the nitrogen-bearing tertiary carbon in 1 was instead of one  $sp^3$  methine group ( $\delta_{\rm H}$  4.81,  $\delta_{\rm C}$  53.4) in the structure of compound 2. Thus, compound 2 was determined to be fumiquinazoline B, which was confirmed by the HMBC correlations (Fig. 3) and comparison with the literature [6].

Compound **3** was isolated as a white solid and optically active  $\left[\alpha\right]_{D}^{25}=-72$  (c 0.07, MeOH). The  $^{1}$ H NMR spectrum of **3** indicated the presence of six methyl groups (two singlet signals and four doublet signals) at  $\delta_{H}$  0.54 (s, CH<sub>3</sub>-18), 0.80 (d, J=7.2 Hz, CH<sub>3</sub>-27), 0.82 (d, J=7.8 Hz, CH<sub>3</sub>-26), 0.88 (d, J=6.6 Hz, CH<sub>3</sub>-28), 0.90 (s, CH<sub>3</sub>-19), 0.99 (d, J=6.6 Hz, CH<sub>3</sub>-21), three olefinic protons at  $\delta_{H}$  5.05 (dd, J=5.4, 2.4 Hz, H-7), 5.17 (dd, J=15.6, 8.4 Hz, H-22), 5.24 (dd, J=15.6, 7.2 Hz, H-23), two oxygenated methine groups at  $\delta_{H}$  3.36 (d, J=5.4 Hz, H-6), 3.77 (1H, m, H-3), and of the overlapped

remaining protons in the aliphatic region at  $\delta_{\rm H}$ 1.43-2.03 (ppm). The  $^{13}$ C NMR and HSQC spectra of compound 3 displayed the signals of 28 carbons, including six methyl groups, seven methylene groups, two oxymethine groups, three  $sp^2$  methine carbons, six  $sp^3$  methine carbons, two quaternary carbons, one oxygen bearing tertiary carbon and one nonprotonated carbon. In addition, a coupling constant of 15.6 Hz between the olefinic protons H-22 and H-23 suggested an E configuration for the double bond. These NMR data indicated that compound 3 possessed a (22*E*)-ergosta-7,22-diene steroid skeleton, characterized by two hydroxyl groups bonded to the carbon at positions C-3 and C-6. Combining the 1D-NMR data, optical rotation value, and comparison with the reported confirmed literature the structure compound 3 as cerevisterol [8].

Compound 4 was isolated as a colorless needle and optically active  $\left[\alpha\right]_{0}^{25}$  = +93.8 (c 0.05, CHCl<sub>3</sub>). The <sup>1</sup>H NMR spectrum of **4** indicated the presence of three methyl groups (one singlet and two doublets) at  $\delta_{\rm H}$  1.31 (d, J=6.6 Hz,  $CH_3$ -13), 1.33 (d, J = 6.6 Hz,  $CH_3$ -12), 2.10 (s,  $CH_3$ -11), two quintet  $sp^3$  methine protons at  $\delta_{\rm H}$  2.97 (dt, J = 7.2, 6.0 Hz, H-4), 4.68 (dt, J = 7.8, 6.6 Hz, H-3), and one singletaromatic proton at  $\delta_{\rm H}$  6.30 (s, H-7). The  $^{13}{\rm C}$ NMR spectrum of 4 indicated the presence of 12 carbons. The <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra, with the aid of the HSQC spectrum, revealed the presence of a penta-substituted benzene ring, one carbon carbonyl at  $\delta_{\mathbb{C}}$  168.6 (C-1), one  $sp^3$  oxymethine carbon at  $\delta_{\rm C}$  80.0 (C-3), one  $sp^3$ methine carbon at  $\delta_{\rm C}$  34.8 (C-4), and three methyl groups. The chemical shifts of the nonprotonated aromatic carbons at  $\delta_{\rm C}$  161.0 (C-6), and 162.4 (C-8) suggested that they were directly linked to the oxygen atoms. Detailed analyses of the 1D-NMR spectra of compound 4 indicated that compound 4 possessed an isochroman-1-one skeleton, which supported by the HMBC correlations as follows: the HMBC correlations from H-7 ( $\delta_{\rm H}$  6.30) to C-5 ( $\delta_{\rm C}$  113.3)/C-6 ( $\delta_{\rm C}$  161.0)/C-8 ( $\delta_{\rm C}$  162.4)/C-9  $(\delta_{\rm C}~100.4)$ , from H-3  $(\delta_{\rm H}~4.68)$  to C-4  $(\delta_{\rm C}$ 34.8)/C-10 ( $\delta_{\rm C}$  143.1)/C-1 ( $\delta_{\rm C}$  168.6), from H-4  $(\delta_{\rm H} \ 2.97)$  to C-5  $(\delta_{\rm C} \ 113.3)$ /C-10  $(\delta_{\rm C} \ 143.1)$ /C-9  $(\delta_{\rm C}~100.4)$  (Fig. 3). Furthermore, the HMBC correlations of the protons H<sub>3</sub>-11 ( $\delta_{\rm H}~2.10$ ) with C-5 ( $\delta_{\rm C}~113.3$ )/C-6 ( $\delta_{\rm C}~161.0$ )/C-10 ( $\delta_{\rm C}~143.1$ ), H<sub>3</sub>-13 ( $\delta_{\rm H}~1.31$ ) with C-4 ( $\delta_{\rm C}~34.8$ )/C-10 ( $\delta_{\rm C}~143.1$ )/C-3 ( $\delta_{\rm C}~80.0$ ), and H<sub>3</sub>-12 ( $\delta_{\rm H}~1.33$ ) with C-3 ( $\delta_{\rm C}~80.0$ )/C-4 ( $\delta_{\rm C}~34.8$ ) confirmed the positions of the methyl groups at C-5, C-4, C-3, respectively (Fig. 3). Therefore, based on the 1D and 2D-NMR data of compound 4 and comparison with the literature, compound 4 was determined to be stoloniferol B [9].

Compound 5 was isolated as a white solid. The <sup>1</sup>H NMR spectrum, with the aid of the COSY spectrum, showed the presence of seven aromatic protons, including one orthodisubstituted benzene ring (from H-5 to H-8), two *ortho*-coupled aromatic protons at  $\delta_{\rm H}$  8.32 (d, J = 5.5 Hz, H-3), 8.11 (d, J = 5.0 Hz, H-4). The <sup>13</sup>C NMR and HSQC spectra of compound **5** displayed the signals of 11 carbons, including seven aromatic methine carbons and four nonprotonated aromatic carbons. The chemical shifts of the non-protonated aromatic carbons at  $\delta_{\rm C}$  142.8 (C-8a), 137.7 (C-9a), and 138.5 (C-3) suggested their linkages to nitrogen atoms. The HMBC spectrum of compound 5 showed the cross-correlations of the protons H-4 ( $\delta_{\rm H}$  8.11) with C-4b ( $\delta_{\rm C}$  122.4)/C-9a ( $\delta_{\rm C}$  137.7), H-3 ( $\delta_{\rm H}$ 8.32) with C-4a ( $\delta_{\rm C}$  130.2)/C-1 ( $\delta_{\rm C}$  134.2), H-1  $(\delta_{\rm H} \, 8.81)$  with C-3  $(\delta_{\rm C} \, 138.5)$ /C-9a  $(\delta_{\rm C} \, 137.7)$ /C-4a ( $\delta_{\rm C}$  130.2), H-5 ( $\delta_{\rm H}$  8.21) with C-8a ( $\delta_{\rm C}$ 142.8)/C-7 ( $\delta_{\rm C}$  129.7), H-6 ( $\delta_{\rm H}$  7.29) with C-4b  $(\delta_{\rm C}\ 122.4)/{\rm C}$ -8  $(\delta_{\rm C}\ 112.9)$ , H-7  $(\delta_{\rm H}\ 7.58)$  with C-8a ( $\delta_{\rm C}$  142.8), which confirmed the positions of C-4, C-3, C-1, C-5, C-6, C-7, respectively. Based on the above evidence and compared with those reported in the reference [10], compound 5 was determined to be norharman.

Compound **6** was isolated as a yellow solid. The  $^1$ H NMR spectrum of compound **6** showed the presence of two singlet aromatic protons at  $\delta_{\rm H}$  7.96 (s, H-5), 7.69 (s, H-8), two singlet methyl groups at  $\delta_{\rm H}$  2.48 (s, CH<sub>3</sub>-10) and 2.50 (s, CH<sub>3</sub>-11). The  $^{13}$ C NMR and DEPT of compound **6** displayed the signals of 12 carbons, including seven non-protonated aromatic carbons, two methyl groups at  $\delta_{\rm C}$  19.2 (CH<sub>3</sub>-10), 20.3 (CH<sub>3</sub>-11), two  $sp^2$  methine carbons at  $\delta_{\rm C}$  125.7 (C-8), 128.5 (C-5), and one carbonyl carbon at  $\delta_{\rm C}$ 

160.6 (COOH). The chemical shifts of four non-protonated aromatic carbons:  $\delta_{\rm C}$  138.6 (C-8a),  $\delta_{\rm C}$  149.8 (C-2),  $\delta_{\rm C}$  146.2 (C-3) and  $\delta_{\rm C}$  141.5 (C-4) suggested their linkages to oxygen or nitrogen atoms. Detailed analyses of 1D NMR data of compound **6** and comparison with the reported data [11] allowed for identifying compound **6** as 3,4-dihydroxy-quinolin-2-carboxylic.

Based on 1D NMR data, comparison with the literature, and co-TLC analysis of authentic samples, the remaining compounds were identified as uracil (7) [12], 3-methyl uracil (8) [13], thymine (9) [12, 13]. Compounds 2 and 8 were first recognized from the genus *Penicillium*, while the remaining compounds were previously isolated from this genus [4, 7, 14–16].

# Antimicrobial assay

Compounds 1-6 were evaluated for their antimicrobial activity against a panel of test microorganisms (three Gram-positive bacteria: Enterococcus faecalis, Staphylococcus aureus, Bacillus cereus; three Gram-negative bacteria: Escherichia coli, Pseudomonas aeruginosa, Salmonella enterica and one yeast strain Candida albicans). The results showed that compound 6 displayed strong antimicrobial activity against two Gram-negative bacteria E. coli and S. enterica with the MIC values of 32 and 64 µg/mL, respectively. Compound 1 exhibited selective inhibition against the Grampositive bacteria E. faecalis with a MIC value of μg/mL. In addition, the remaining compounds 2-5 had inhibitory activity against one to three Gram-positive and Gram-negative tested strains, with MIC values ranging from 128 to 256 μg/mL (Table 1).

Table 1. Antimicrobial activities of compounds 1-6 (MIC: μg/mL)

Compd.	Gram-positive			Gram-negative			Yeast
	E. faecalis	S. aureus	B. cereus	E. coli	P. aeruginosa	S. enterica	C. albicans
1	64	128	256	256	> 256	256	128
2	256	> 256	> 256	> 256	> 256	256	128
3	256	256	128	256	> 256	128	128
4	256	256	128	> 256	> 256	> 256	258
5	> 256	> 256	> 256	128	> 256	> 256	256
6	> 256	> 256	> 256	32	> 256	64	> 256
Streptomycin	256	256	128	32	256	128	-
Nystatin	-	-	-	-	-	-	8

Note: "-": not test.

# CONCLUSION

From the agar-based culture of the marinederived Penicillium sp. M839 strain, nine known compounds were isolated and structurally determined, including fumiquinazoline D (1), В cerevisterol fumiquinazoline (2), stoloniferol B (4), norhaman (5), 3,4-dihydroxy-6,7-dimethyl-quinolin-2-carboxylic (6), uracil (7), 3-methyl uracil (8), thymine (9). These compounds were characterized via 1D and 2D NMR spectroscopic and mass spectrometric analyses. Compounds 2 and 8 were first recognized from the genus *Penicillium*, while the remaining compounds were previously isolated from this genus. Compounds **1–6** were evaluated for their antimicrobial activity against a panel of test microorganisms (three Grambacteria: Enterococcus faecalis. positive Staphylococcus aureus, Bacillus cereus; three Gram-negative bacteria: Escherichia Pseudomonas aeruginosa, Salmonella enterica and one yeast strain Candida albicans). The results showed that compound 6 displayed strong antimicrobial activity against two Gramnegative bacteria, E. coli and S. enterica, with the MIC values of 32 and 64 µg/mL, respectively. Compound 1 showed antimicrobial activity against all three Gram-positive and two Gramnegative strains (S. enterica, C. albicans) with MIC values from 64 to 256 μg/mL. In addition, the remaining compounds 2-5 had inhibitory activity against one to three Gram-positive and Gram-negative tested strains with MIC values ranging from 128 to 256  $\mu$ g/mL.

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